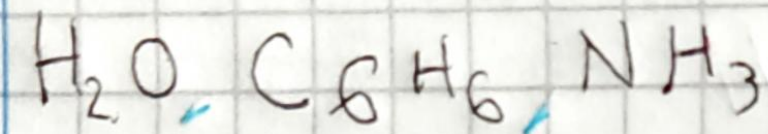
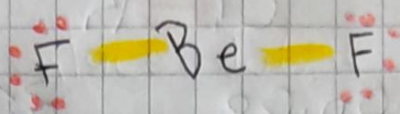
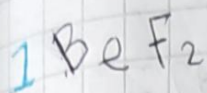


- enlaces simples: F₂, C₂H₄, ~~C₂H₂~~



- enlaces dobles: CS₂

- triples enlace Ninguno



Iônico

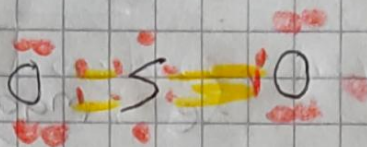
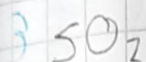
$$3,98 - 7,57 =$$

$$2,41$$



Por ligação co

covalente não polar

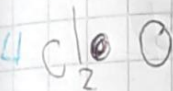


$$3,44 - 2,58 =$$

Covalente

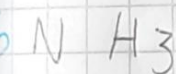
polar

$$0,86$$



Covalente
não polar

$$3,44 - 3,16 = 0,28$$



Covalente
polar

$$3,4 - 2,2 = 1,2$$

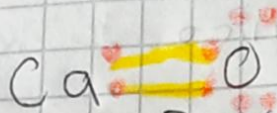
6 HCl



ionico

$$3,16 - 0,82 = 2,34$$

7 CaO



ionico

$$3,44 - 1,27 = 2,17$$

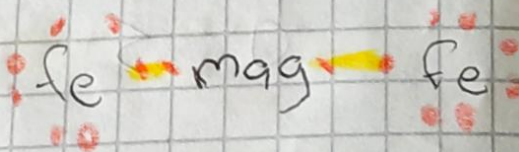
8 NO



covalente no polar

$$3,44 - 3,04 = 0,4$$

9 mag Fe₂



covalente

$$1,83 - 1,31$$

$$= 0,52$$

10 HCl



covalente polar

$$3,16 - 2,2 = 0,96$$

