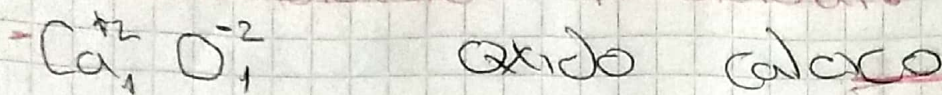
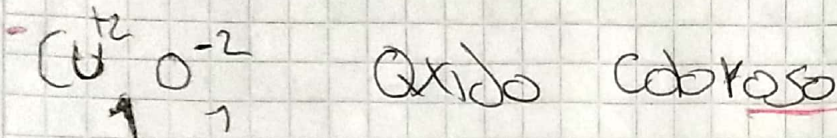


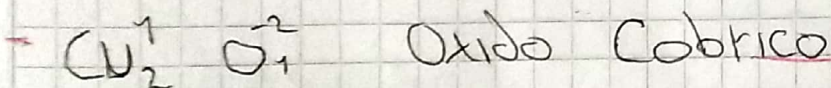
Nomenclatura tradicional:



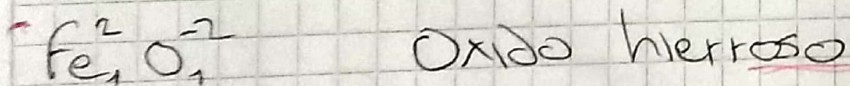
$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



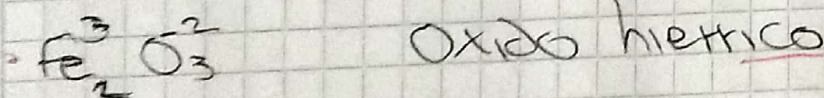
$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



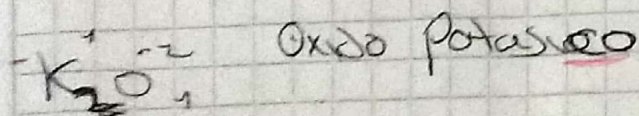
$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



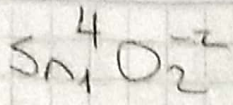
$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



$$\begin{array}{r} -6 \\ +6 \\ \hline 0 \end{array}$$

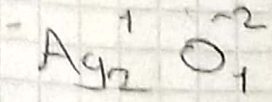


$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



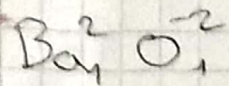
Oxido estánico

$$\begin{array}{r} -4 \\ +4 \\ \hline 0 \end{array}$$



Oxido platínico

$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



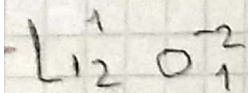
Oxido barico

$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



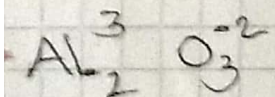
Oxido sódico

$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



Oxido litico

$$\begin{array}{r} -2 \\ +2 \\ \hline 0 \end{array}$$



Oxido aluminico

$$\begin{array}{r} -6 \\ +6 \\ \hline 0 \end{array}$$

Sistemática

- CaO = ^{monóxido} Óxido de calcio.
- Cu_2O = monóxido de dicobre.
- CuO = ^{monóxido} Óxido de cobre.
- FeO = óxido de hierro.
- Fe_2O_3 = trióxido de di hierro.
- K_2O = monóxido de dipotasio.
- SnO_2 = dióxido de estánico.
- Ag_2O = monóxido de diplata.
- BaO = monóxido de bario.
- Na_2O = monóxido de disodio.
- Li_2O = monóxido de dilítico.
- Al_2O_3 = trióxido de dialuminio.

Stock

- CaO = Óxido de Calcio (II)
- Cu_2O = Óxido de Cobre (I)
- CuO = Óxido de Cobre (II)
- FeO = Óxido de hierro (II)
- Fe_2O_3 = Óxido de hierro (III)
- K_2O = Óxido de potasio.
- SnO_2 = Óxido de estaño (IV)
- Ag_2O = Óxido de Plata.
- BaO = Óxido de bario (II)

Na_2O = Oxido de Sodio.

Li_2O = Oxido de litio.

Al_2O_3 = Oxido de aluminio (III)