- $CaCO_2$

$Ca = 41 \times 1 = 41$
$CO = 56 \times 2 = 112$
$= 155 = UMO$

- $Fe(NO_3)_3$

$Fe = 56 \times 1 = 56$
$N = 14 \times 3 = 42$
$O = 16 \times 4 = 144$
$= 242 = UMO$

- C_1C_1

$Cl = 1 \times 1 = 1$
$C = 1 \times 12 = 12$
$I = 1 \times 127 = 127$
$= 140 = UMO$

- $Al(O_4)$

$Al = 27 \quad 1 \times 1$
$O = 16 \quad 2 \times 16$
$4 = 1 \quad 1 \times 2$
$= 35 = UMO$

- O_4NO_3

$4 = 1 \quad 1 \times 1$
$N = 14 \quad 1 \times 14$
$O = 16 \quad 1 \times 16$
$= 63 = UMO$

- H_2SO_4

$H_2 = 1 \quad 2 \times 1 = 2$
$S = 32 \quad 1 \times 32 = 32$
$O_4 = 16 \quad 4 \times 16 = 64$
$= 97 \ UMO$

- $C_8H_{12}C_6$

  C = 12 → 8 × 12 = 96
  H = 1 → 12 × 1 = 12
  O = 16 → 6 × 16 = 96
  = 204 UMO

- $NaOH$

  Na = 1 × 23 = 23
  O = 16 → 4 × 16 = 64
  H = 1 → 1 × 1 = 1
  = 88 UMO

- MgO

  Mg = 24 → 2 × 24 = 40 UMO
  O = 16 → 1 × 16

- $CuSO_4$

  Cu = 1 × 63 = 63
  S = 2 × 32 = 34
  O = 4 × 16 = 64
  261 UMO

- NH_3

  N = 14 → 1 × 14 = 14
  H = 1 → 3 × 1 = 3
  = 17 UMO

- C_6H_{14}

  C = 12 → 6 × 12 = 72
  H = 1 → 14 × 1 = 14
  86 UMO

- C_2H_2

  C = 12 → 1 × 12 = 24
  H = 16 → 2 × 16 = 32
  = 56 UMO

- CO_2

  C = 12 → 1 × 12 = 24
  O = 16 → 2 × 16 = 32
  = 56 UMO

- FeO_2

  Fe = 56 → 2 × 56 = 112
  O = 16 → 2 × 16 = 32
  = 144 UMO