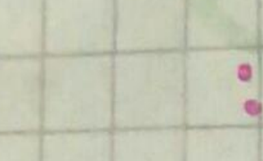
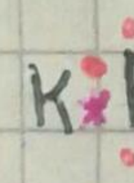
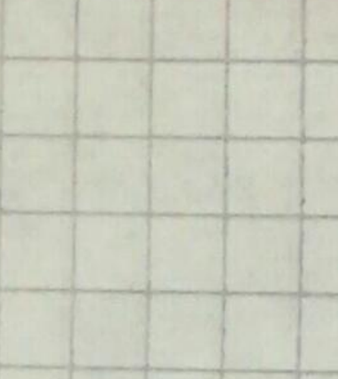
BaF_2 Ba(metal) F(no metal) $BaF_2 = 3,98 - 0,89 = 3,09$ Iónico



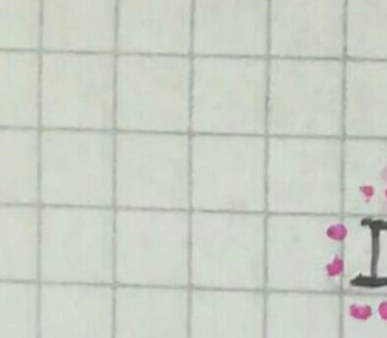
KBr K(metal) Br(no metal) $KBr = 2.96 - 0.82 = 2.14$ Iónico



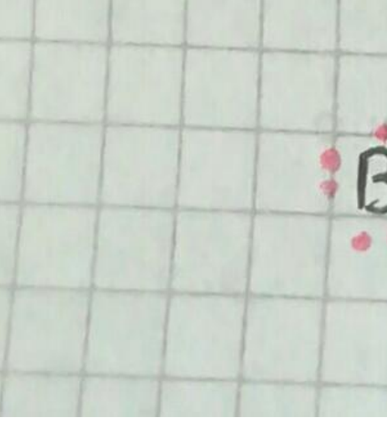
$FeCl_3$ Fe(metal) Cl(no metal) $Fe-Cl = 3.16 - 1,83 = 1,33$ Iónico



AlI_3 Al(metal) I(no metal) $Al-I = 2,66 - 1,61 = 1,05$ Covalente polar



$CaBr_2$ Ca(metal) Br(no metal) $Ca-Br = 2,96 - 1,00 = 1,96$ Iónico



Scanned by TapScanner