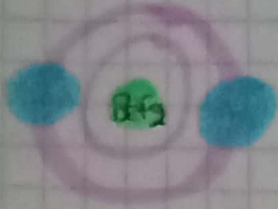
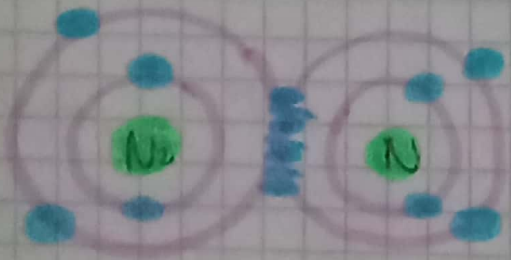


Taller

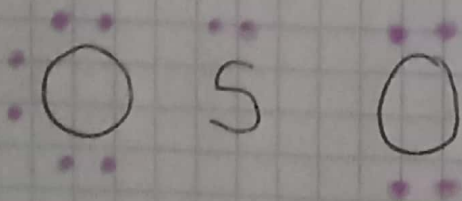
a) BeF_2 —▶ Enlace covalente

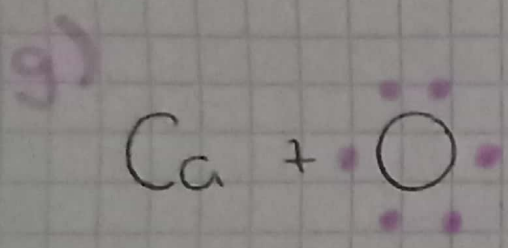
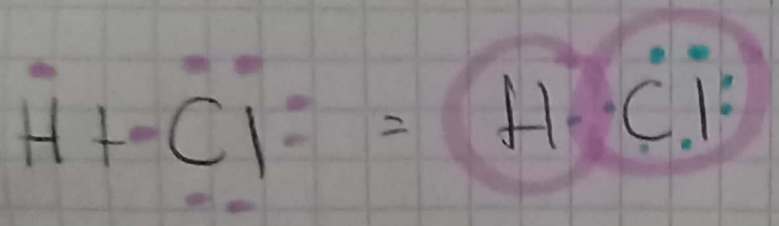
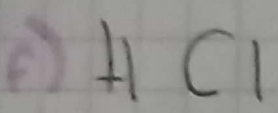
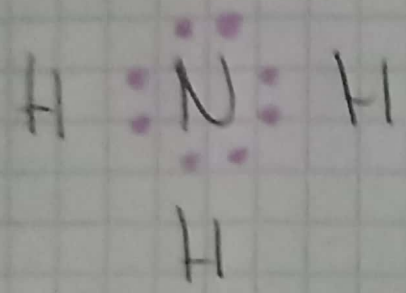
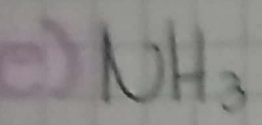
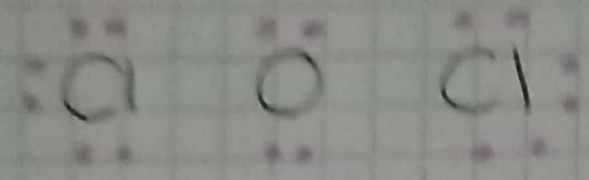
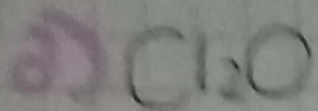


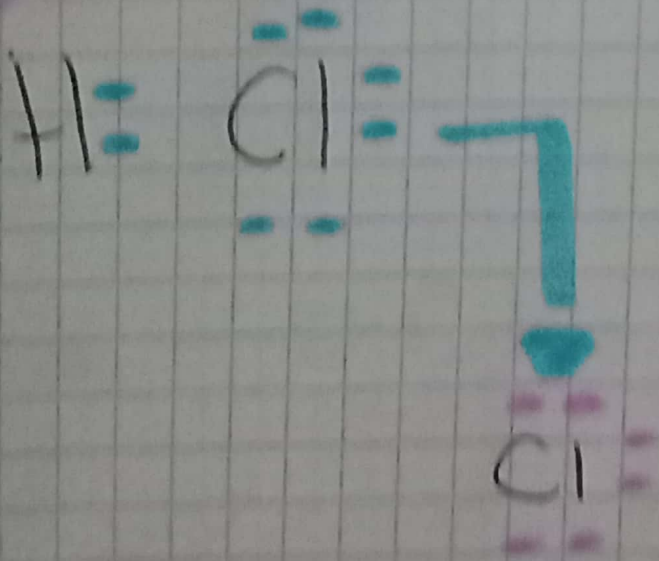
b) N_2 —▶ Enlace Covalente



c) SO_2

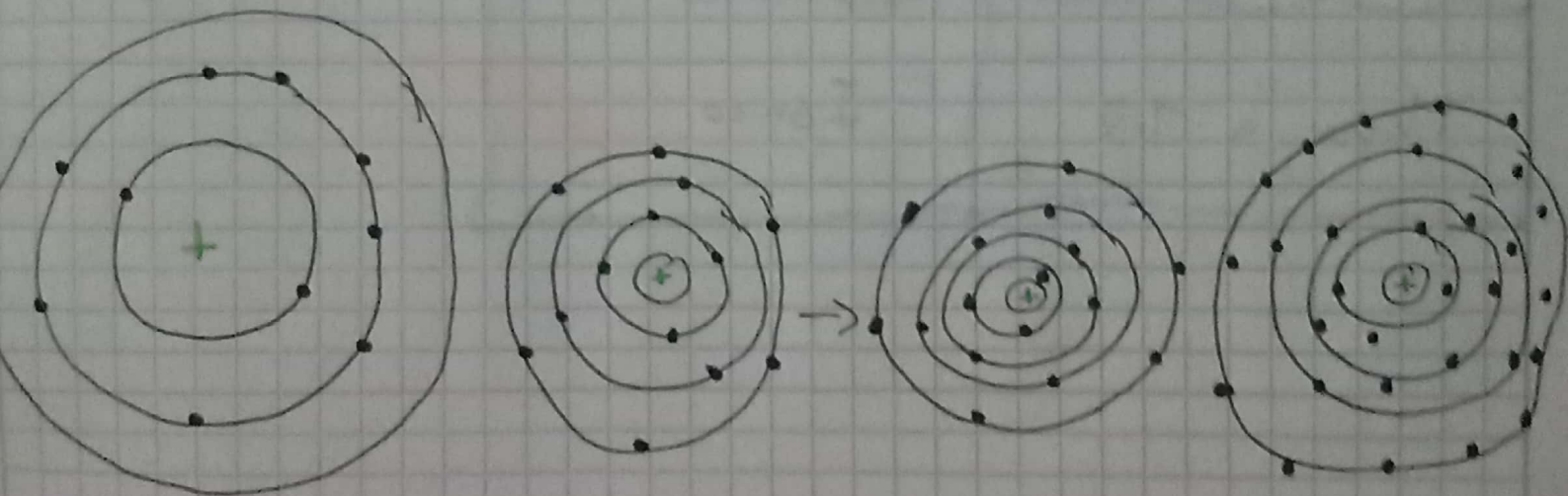
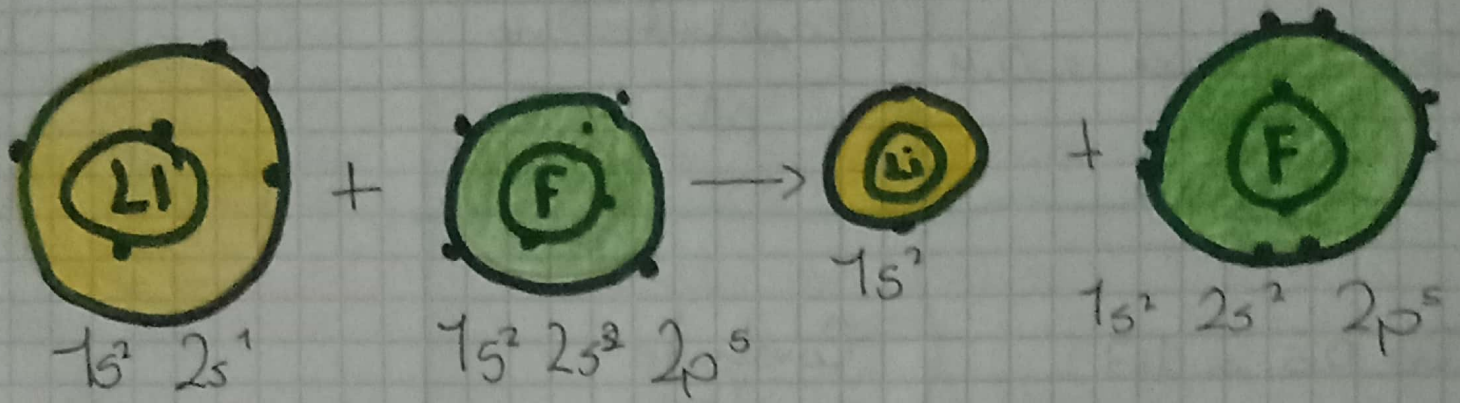






Enlaces

iónicos



Ejercicios

$$\text{NaCl} = \text{Cloruro de sodio} = 3.16 - 0.93 = 2.23$$

$$\text{BaCl}_2 = \text{Cloruro de bario} = 3.16 - 0.89 = 2.27$$

$$\text{Na}_2\text{S} = \text{Sulfuro de sodio} = 2.58 - 0.93 = 1.65$$

$$\text{BaS} = \text{Sulfuro de bario} = 2.58 - 0.89 = 1.69$$