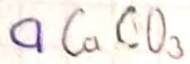


Ejercicios

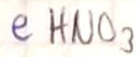


$$\text{Ca} = 40 \times 1 = 40$$

$$\text{C} = 12 \times 1 = 12$$

$$\text{O} = 16 \times 3 = 48$$

100 uma

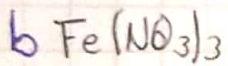


$$\text{H} = 1 \times 1 = 1$$

$$\text{N} = 14 \times 1 = 14$$

$$\text{O} = 16 \times 3 = 48$$

63 uma

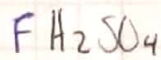


$$\text{Fe} = 56 \times 1 = 56$$

$$\text{N} = 14 \times 3 = 42$$

$$\text{O} = 16 \times 9 = 144$$

242 uma

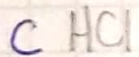


$$\text{H} = 1 \times 2 = 2$$

$$\text{S} = 32 \times 1 = 32$$

$$\text{O} = 16 \times 4 = 64$$

98 uma

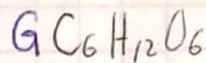


$$\text{H} = 1 \times 1 = 1$$

$$\text{C} = 12 \times 1 = 12$$

$$\text{I} = 127 \times 1 = 127$$

140 uma

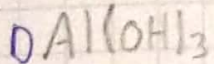


$$\text{C} = 12 \times 6 = 72$$

$$\text{H} = 1 \times 12 = 12$$

$$\text{O} = 16 \times 6 = 96$$

180 uma

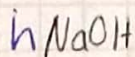


$$\text{Al} = 27 \times 1 = 27$$

$$\text{O} = 16 \times 3 = 48$$

$$\text{H} = 1 \times 3 = 3$$

78 uma



$$\text{Na} = 23 \times 1 = 23$$

$$\text{O} = 16 \times 1 = 16$$

$$\text{H} = 1 \times 1 = 1$$

40 uma

I MgO

$$\text{Mg} = 24 \times 1 = 24$$

$$\text{O} = 16 \times 1 = 16$$

40 uma

M C₂H₂

$$\text{C} = 12 \times 2 = 24$$

$$\text{H} = 1 \times 2 = 2$$

26 uma

J CuSO₄

$$\text{Cu} = 63 \times 1 = 63$$

$$\text{S} = 32 \times 1 = 32$$

$$\text{O} = 16 \times 4 = 64$$

159 uma

N CO₂

$$\text{C} = 12 \times 1 = 12$$

$$\text{O} = 16 \times 2 = 32$$

44 uma

K NH₃

$$\text{N} = 14 \times 1 = 14$$

$$\text{H} = 1 \times 3 = 3$$

17 uma

O Fe₂O₃

$$\text{Fe} = 56 \times 2 = 112$$

$$\text{O} = 16 \times 3 = 48$$

160 uma

L C₆H₁₄

$$\text{C} = 12 \times 6 = 72$$

$$\text{H} = 1 \times 14 = 14$$

86 uma