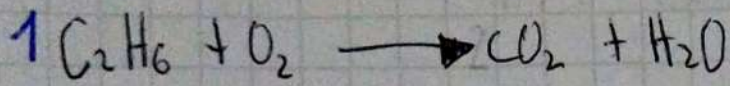


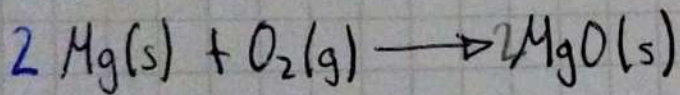
Aktividad



$$2 = \text{C} = 2$$

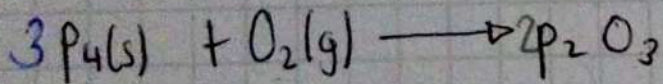
$$6 = \text{H} = 6$$

$$8 = \text{O} = 8$$



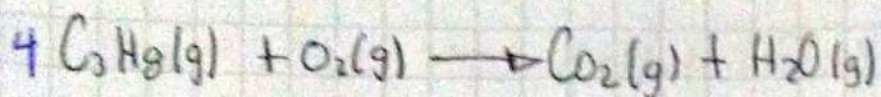
$$1 = \text{Mg} = 1$$

$$2 = \text{O} = 2$$



$$4 = \text{P} = 4$$

$$3 = \text{O} = 3$$



$$= \text{C} =$$

$$= \text{H} =$$

$$= \text{O} =$$



$$2 = \text{Cr} = 2$$

$$6 = \text{Cl} = 6$$

$$6 = \text{H} = 6$$



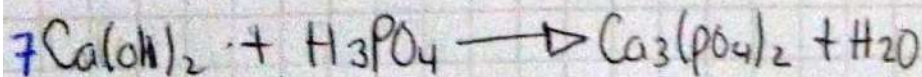
$$2 = \text{N} = 2$$

$$6 = \text{H} = 6$$

$$2 = \text{Cl} = 2$$

$$2 = \text{Ba} = 2$$

$$1 = \text{O} = 1$$



$$3 = \text{Ca} = 3$$

$$6 = \text{O} = 6$$

$$6 = \text{P} = 6$$

$$6 = \text{H} = 6$$



$$1 = \text{Zn} = 1$$

$$1 = \text{S} = 1$$

$$2 = \text{Cl} = 2$$

$$2 = \text{H} = 2$$



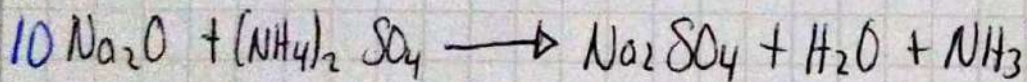
$$2 = \text{Fe} = 2$$

$$6 = \text{Cl} = 6$$

$$6 = \text{Na} = 6$$

$$3 = \text{C} = 3$$

$$3 = \text{O} = 3$$



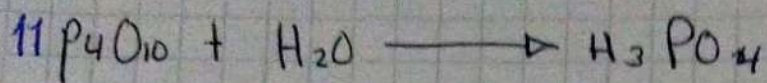
$$2 = \text{Na} = 2$$

$$2 = \text{N} = 2$$

$$6 = \text{H} = 6$$

$$1 = \text{S} = 1$$

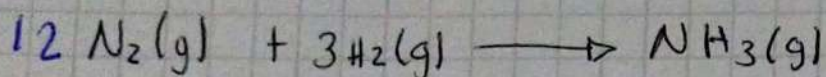
$$4 = \text{O} = 4$$



$$4 = \text{P} = 4$$

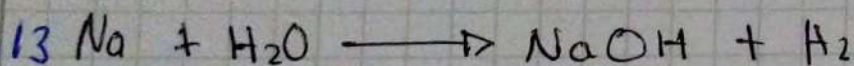
$$7 = \text{H} = 7$$

$$10 = \text{O} = 10$$



$$2 = \text{N} = 2$$

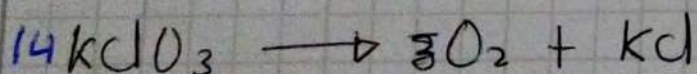
$$5 = \text{H} = 5$$



$$2 = \text{Na} = 2$$

$$4 = \text{H} = 4$$

$$2 = \text{O} = 2$$



$$2 = \text{K} = 2$$

$$2 = \text{Cl} = 2$$

$$5 = \text{O} = 5$$