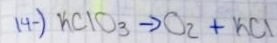
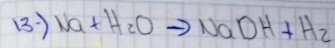


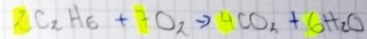
Balances de Ecuaciones Químicas

- 1) $C_2H_6 + O_2 \rightarrow CO_2 + H_2O$
- 2) $Mg (s) + O_2 (g) \rightarrow MgO (s)$
- 3) $P_4 (s) + O_2 (g) \rightarrow P_2O_5 (g)$
- 4) $C_3H_8 (g) + O_2 (g) \rightarrow CO_2 (g) + H_2O (g)$
- 5) $HCl (ac) + Cr (s) \rightarrow CrCl_3 (ac) + H_2 (g)$
- 6) $NH_4Cl + Ba(OH)_2 \rightarrow BaCl_2 + NH_3 + H_2O$
- 7) $Ca(OH)_2 + H_3PO_4 \rightarrow Ca_3(PO_4)_2 + H_2O$
- 8) $ZnS + HCl \rightarrow ZnCl_2 + H_2S$
- 9) $FeCl_3 + Na_2CO_3 \rightarrow Fe_2(CO_3)_3 + NaCl$
- 10) $Na_2O + (NH_4)_2SO_4 \rightarrow Na_2SO_4 + H_2O + NH_3$
- 11) $P_4O_{10} + H_2O \rightarrow H_3PO_4$
- 12) $N_2 (g) + H_2 (g) \rightarrow NH_3 (g)$

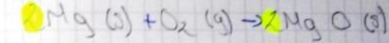


Solución

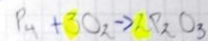
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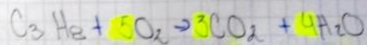
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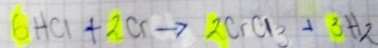
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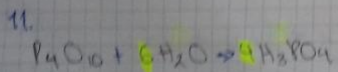
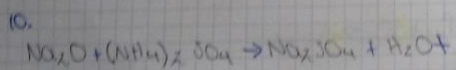
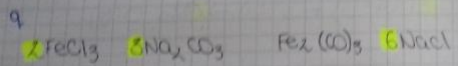
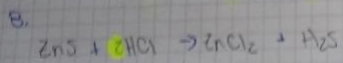
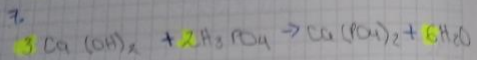
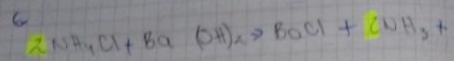


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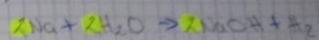
5.





12.

13.



14.

