

Obten la normalidad de ^{65.5} $\text{Al}(\text{OH})_3$ con 450ml y 9.50 gramos

Datos:

$$m: 65.5\text{g}$$
$$V: 450\text{ml} / 1000 = 0.45$$

$$\text{PM: Al} = 1 \times 27$$
$$H = 3 \times 1$$
$$O = 3 \times 16$$
$$44\text{g/mol}$$

$$Eg: 3$$

$$N = \frac{65.5 \times 3}{44 \times 0.45}$$

$$N = \frac{196.5}{49.5}$$

$$N = 3.97$$