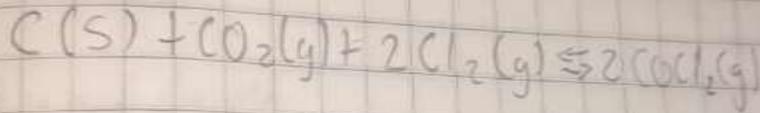
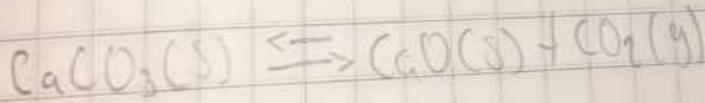


ejercicios



$$K_c = \frac{[\text{COCl}_2]^2}{[\text{CO}_2][\text{Cl}_2]^2} = \frac{0,14^2}{0,012 \times 0,054^2} = 220$$



$$K_c = [\text{CO}_2] = \frac{0,105}{(0,082 \times 623)^{1-1}} = 2,06 \cdot 10^{-3}$$