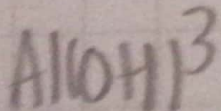


# Actividad

Obten la normalidad de  $\text{Al}(\text{OH})_3$  con 450ml y 9.50 gramos

$$V = 450 \text{ ml}$$

$$m = 9.50 \text{ g}$$



$$m = 9.50 \text{ g}$$

$$PM = 78 \text{ g/mol}$$

$$V = 0.45 \text{ L}$$

$$N = ?$$

$$\text{Al} = 27 \times 1 = 27$$

$$\text{O} = 16 \times 3 = 48$$

$$\text{H} = 1 \times 3 = 3$$

$$N = \frac{(9.50) \cdot (3)}{(78) \cdot (0.45)} = \frac{28.5}{35.1} =$$

$$N = 1.23$$