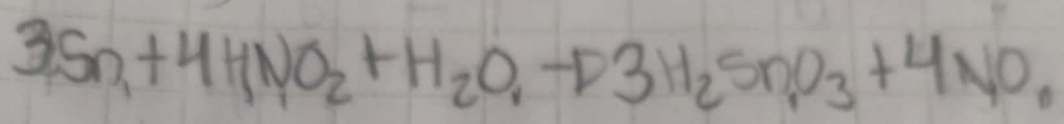
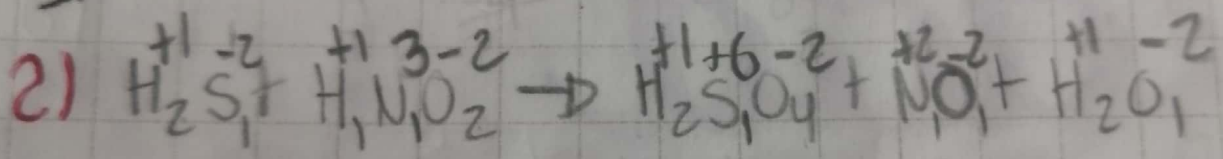


$\overset{+5}{\text{N}} \rightarrow \overset{+2}{\text{N}}$ (3) reduce
 $\overset{0}{\text{Sn}} \rightarrow \overset{+4}{\text{Sn}}$ (4) oxido

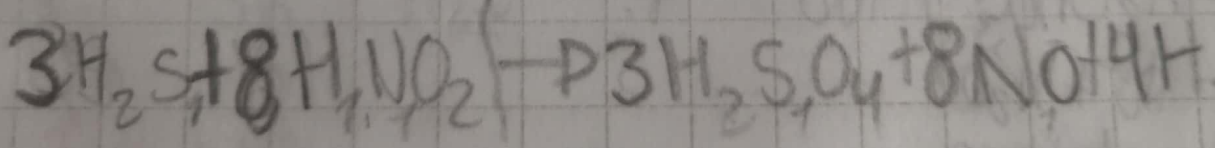
4 N \rightarrow N 4
 3 Sn \rightarrow Sn 3



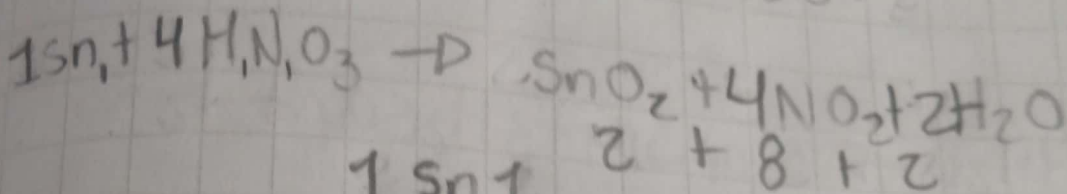
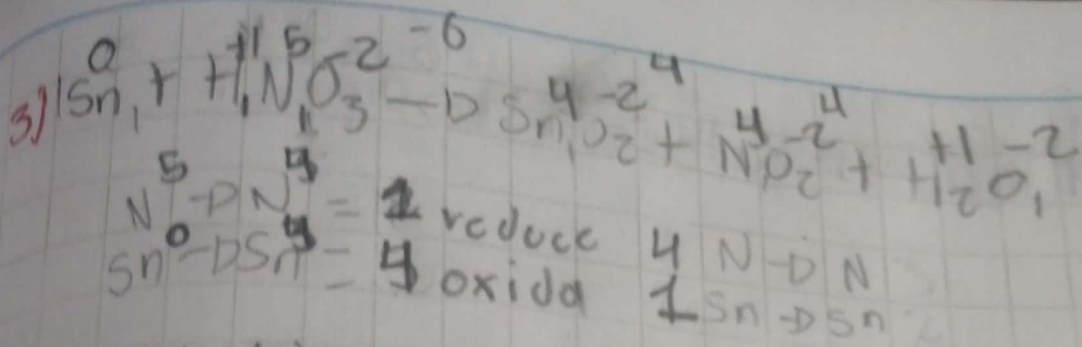
3 Sn 3
 4 N 4
 6 H 6
 13 O 13



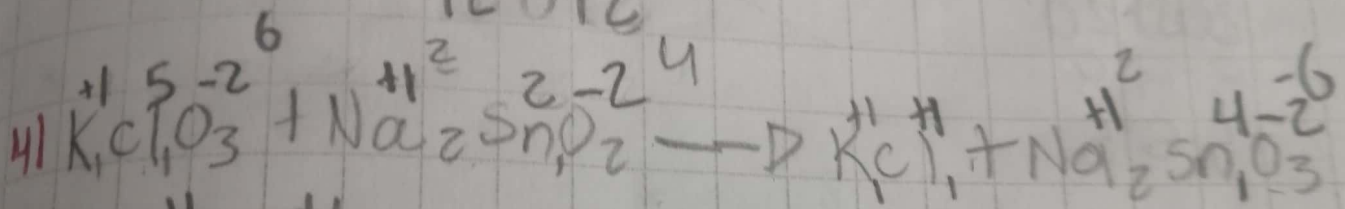
$\overset{+3}{\text{N}} \rightarrow \overset{+1}{\text{N}}$ (3) reduce
 $\overset{-2}{\text{S}} \rightarrow \overset{+6}{\text{S}}$ (8) oxido
 8 N \rightarrow N 8
 3 S \rightarrow S 3



4 6
 3 S 3
 8 N 8
 14 H 14
 24 O 24

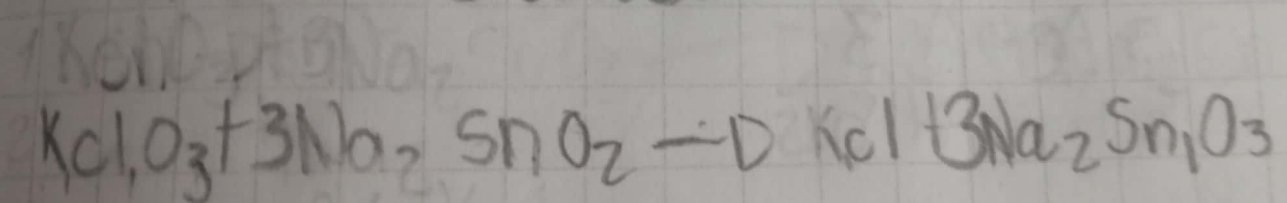


1 Sn 1
 4 N 4
 4 H 4
 12 O 12

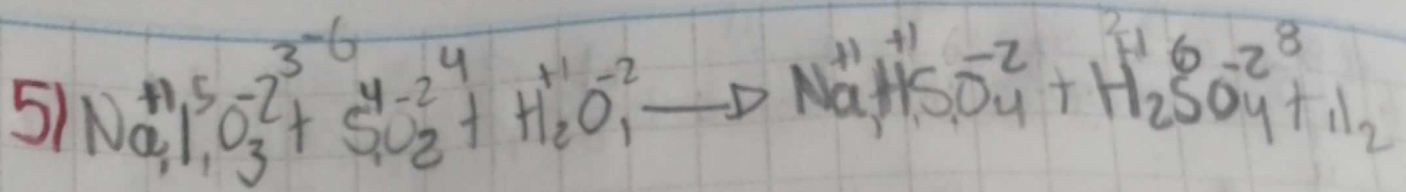


$\overset{+5}{\text{Cl}} \rightarrow \overset{+1}{\text{Cl}} = 4 \text{ reduce}$
 $\overset{+2}{\text{Na}} \rightarrow \overset{+2}{\text{Na}} = 0$
 $\overset{-2}{\text{Sn}} \rightarrow \overset{-4}{\text{Sn}} = 3 \text{ oxida}$

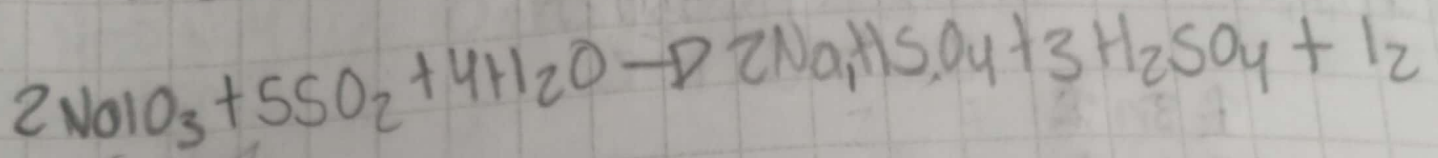
3 Cl \rightarrow Cl 3
 1 Sn \rightarrow Sn 1



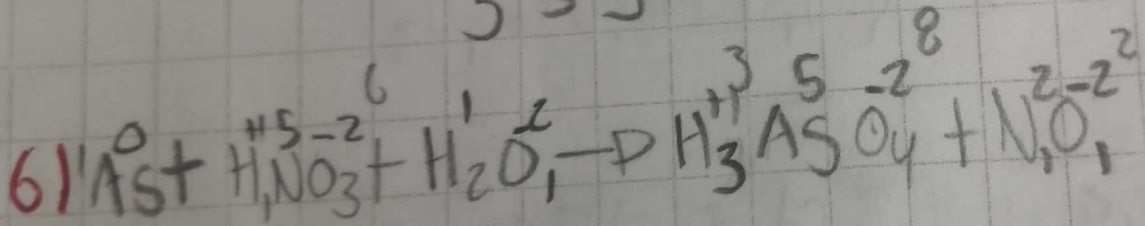
1 K 1
 1 Cl 1
 6 Na 6
 3 Sn 3
 9 O 9



Na → No
 I⁵ → I¹
 S⁴ → S⁶ 6 12

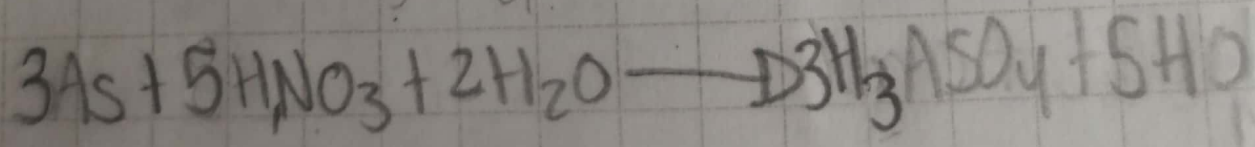


2 Na 2
 2 I 2
 20 O 20
 8 H 8
 5 S 5

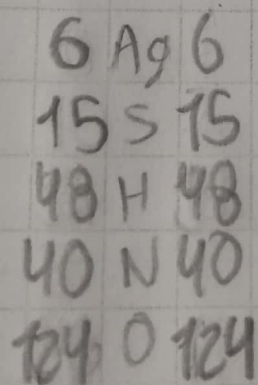
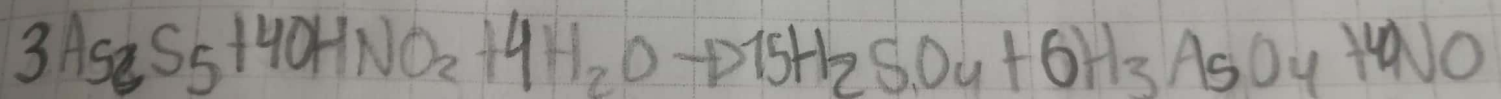
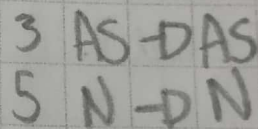
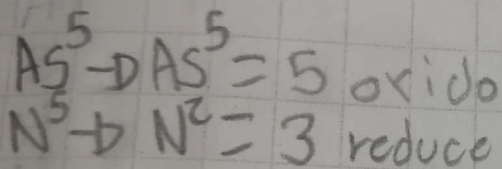
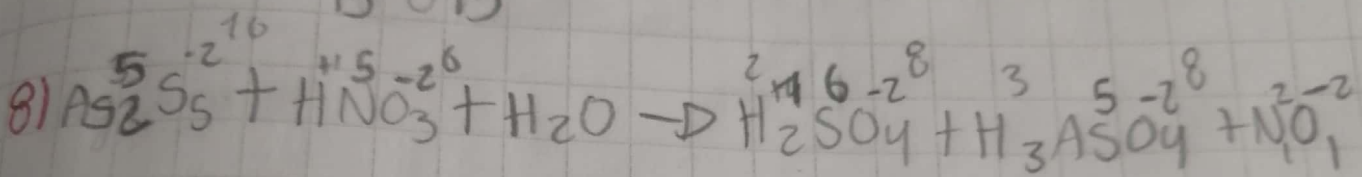
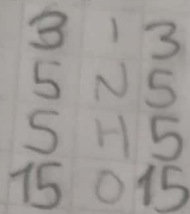
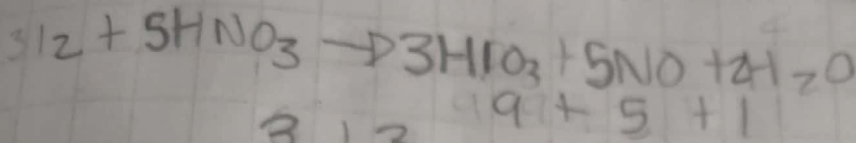
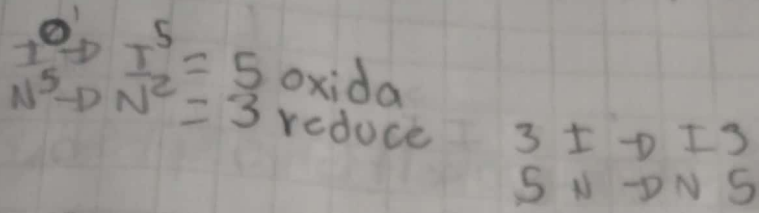
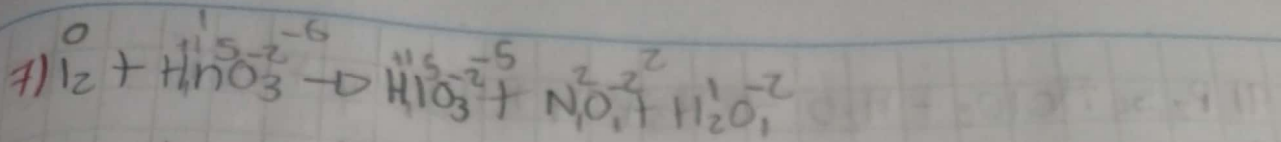


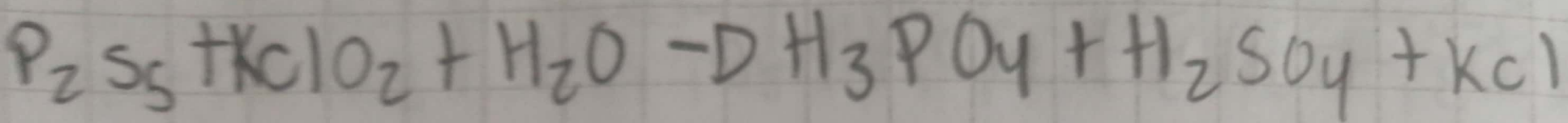
As⁰ → As⁵ - 5 kida
 N⁵ → N² = 3 reduce

3 As → As₃
 5 N → N₅



5 4 3 As 3
 5 N 5
 9 H 9
 17 O 17





15 S reducido
20 Cl oxidado

