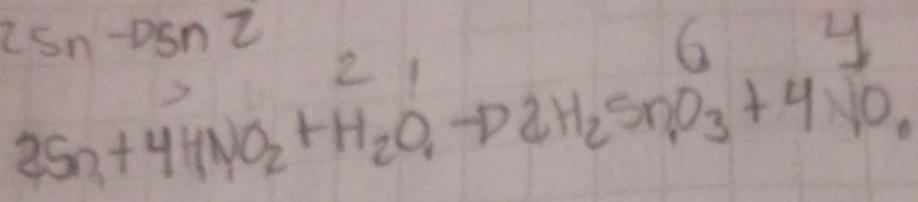
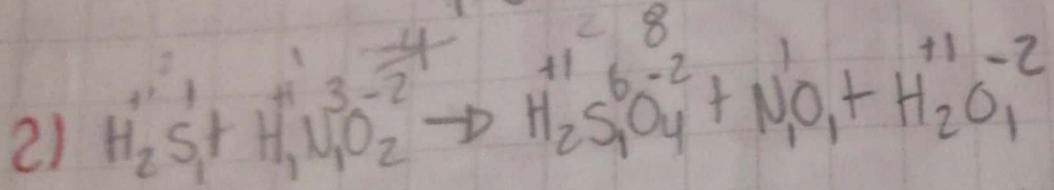


$\overset{0}{\text{Sn}} \rightarrow \overset{4}{\text{Sn}}$  (4) oxidado  
 $\overset{3}{\text{N}} \rightarrow \overset{1}{\text{N}}$  (2) reduce

$4 \text{N} \rightarrow \text{N} 4$   
 $2 \text{Sn} \rightarrow \text{Sn} 2$

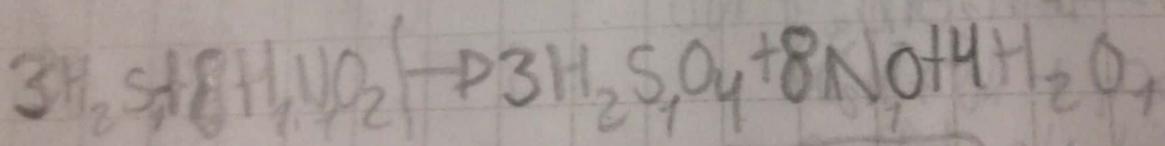


2 Sn 2  
 4 N 4  
 6 H 4  
 9 O 10



$\overset{3}{\text{N}} \rightarrow \overset{1}{\text{N}}$  (2) reduce  
 $\overset{2}{\text{S}} \rightarrow \overset{6}{\text{S}}$  (6) oxidado

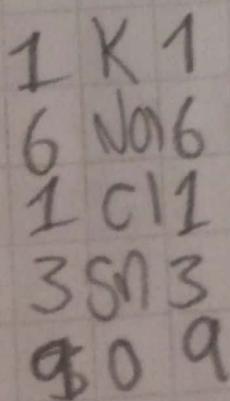
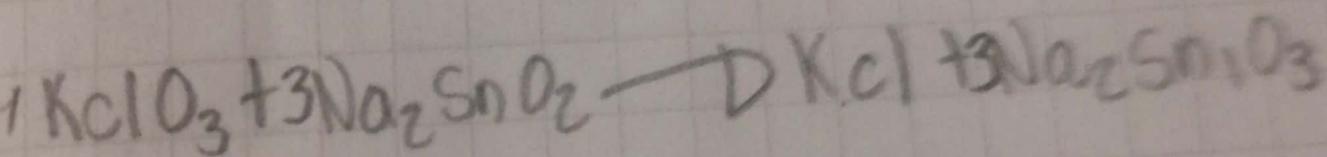
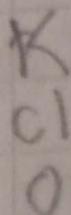
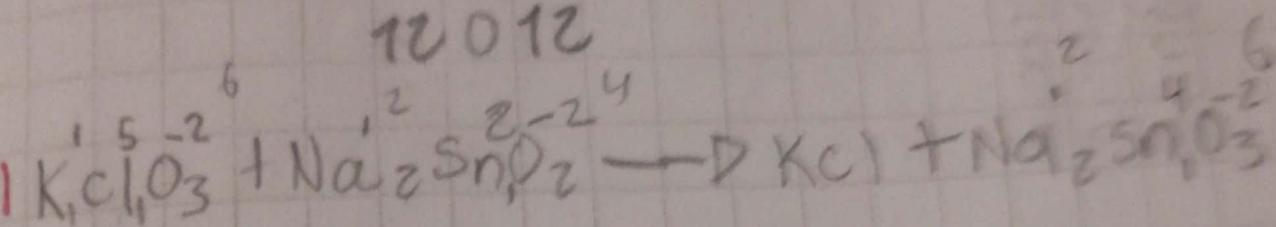
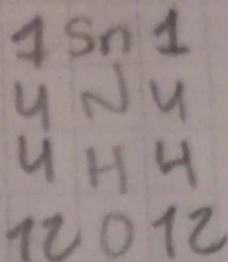
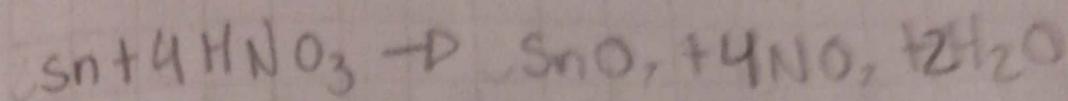
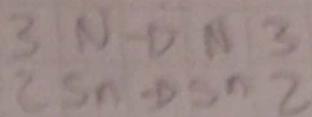
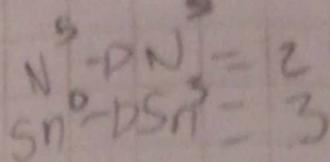
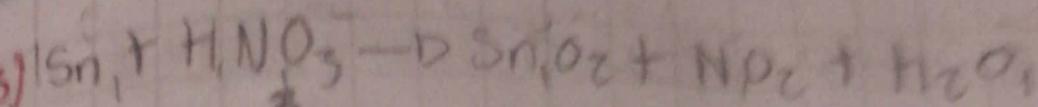
$6 \text{N} \rightarrow \text{N} 6$   
 $2 \text{S} \rightarrow \text{S} 2$

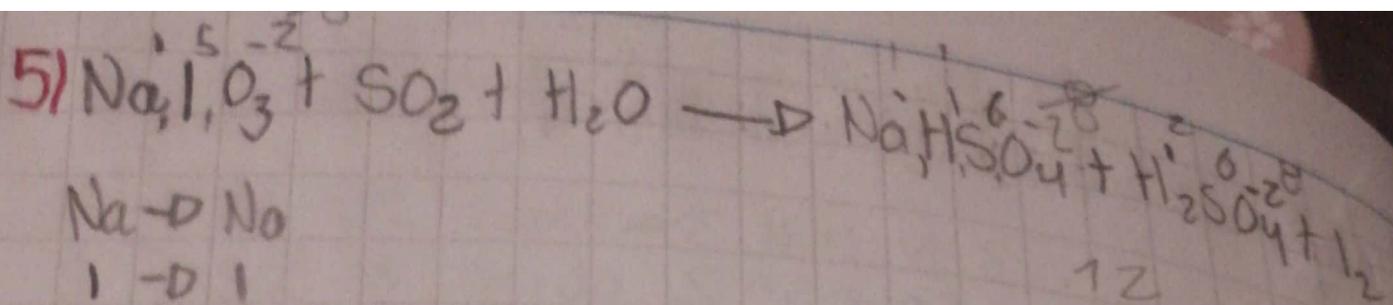


9 S 3  
 8 N 8  
 3 S 3  
 24 O 24  
 14 H 14

4 N 4  
 4 S 4  
 12 O 22  
 12 H 12

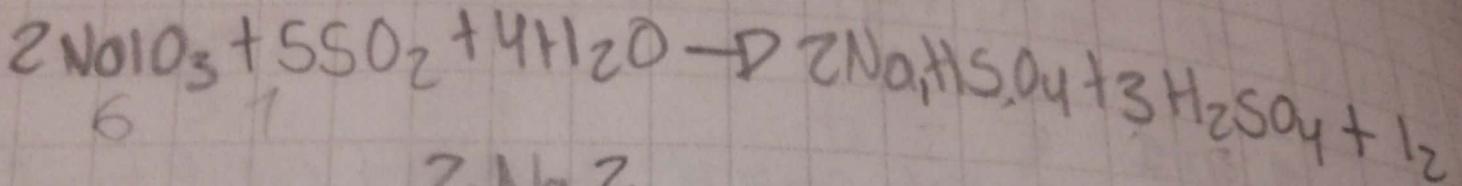
Possible solution



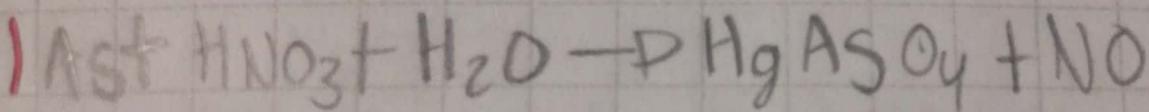


Na → Na  
 I → I  
 S → S 6 12

12

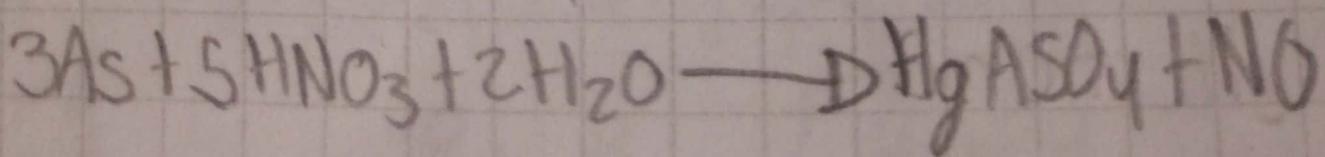


6 1  
 2 Na 2  
 2 I 2  
 20 O 20  
 8 H 8  
 5 S 5

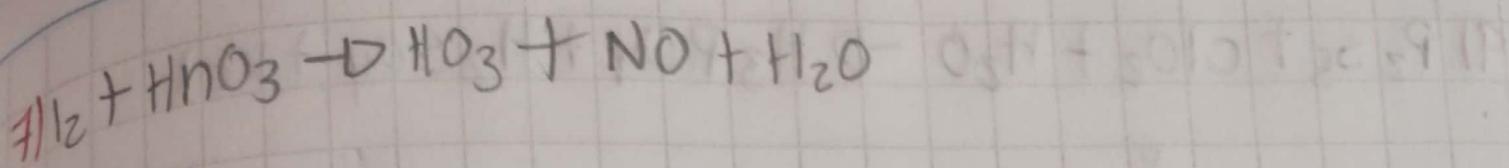


As  
 H  
 N  
 O

5 4

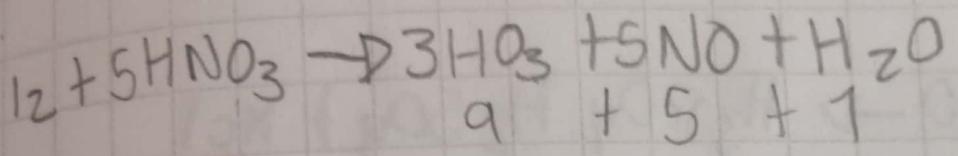


3 As  
 15 N  
 9 H

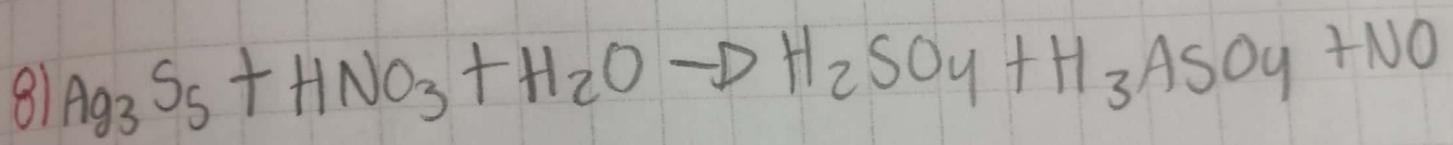


I  
H  
N  
O

3 + 2

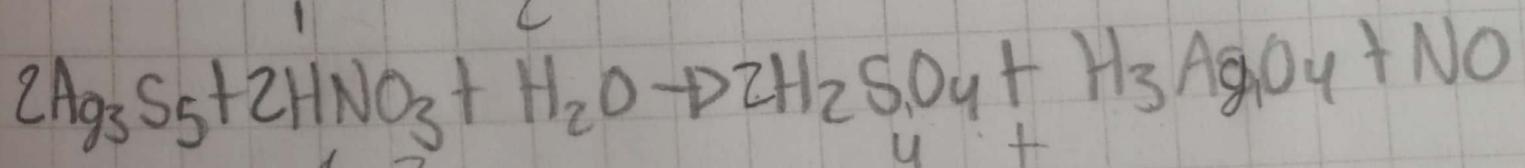


2 1 2  
5 2 5  
5 1 5  
15 0 15



Ag  
S  
H  
N  
O

1 2



1 2

6 Ag  
10 S 2  
4 H 4  
2 N  
7 O 8

4 +  
8 +