

isoelectrónica

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Completa la siguiente tabla de acuerdo a los conceptos estudiados de la tabla periódica.

Elemento	Símbolo	Z	A	P+	e-	N	Metal	No metal	Representativo	Transición
Cromo	Cr	24	52	24	24	52	X			
Azufre	S	16	32	16	16	32		X		
Cobre	Cu	29	63	29	29	63	X	X		
Bismuto	Bi	83	209	83	83	209			X	
Selenio	Se	34	79	34	34	79		X		
Molibdeno	Mo	42	96	42	42	96	X			
Francio	Fr	87	223	87	87	223	X			



17 Realizar la distribución electrónica para $Z=54$, $Z=46$. Hallar periodo y grupo, distribución por niveles ubicando electrones de valencia, fórmula de Lewis del elemento.

$$Z=54 \left\{ \text{Xe} = \text{Xenón} = 6 = 18 \mid P = 5.$$

$$Z = 1s^2, 2s^2, 2p^6, 3s^2, 3p^6, 4s^2, 3d^{10}, 4p^6, 5s^2, 4d^{10}, 5p^6$$

$$Z = 46$$

$$Z =$$

Rejilla de respuestas

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b	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input checked="" type="radio"/>	<input checked="" type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>
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	11	12	13	14	15	16	17	18	19	
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