

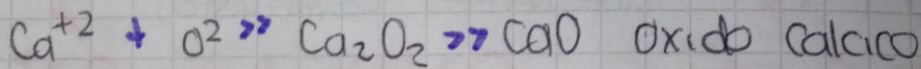
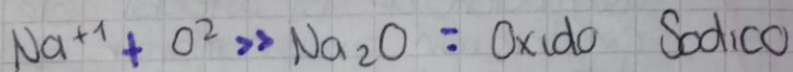
PROPOSITO

Identificar las Características de las nomenclaturas para nombrar a los Oxidos.

OXIDOS BASICOS

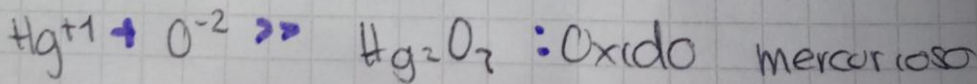
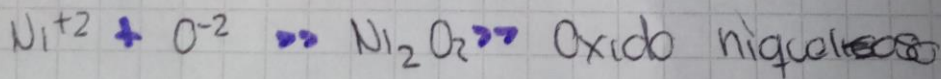
Nomenclatura tradicional

Una Valencia Oxido...ico

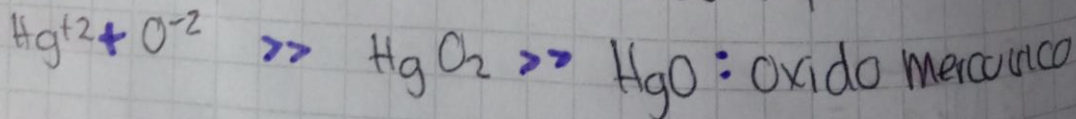
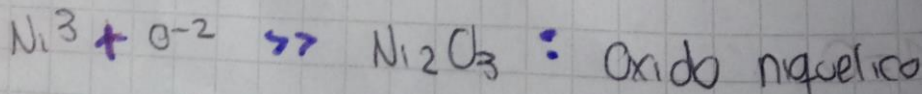


Das Valencias

menor Valencia Oxido...oso

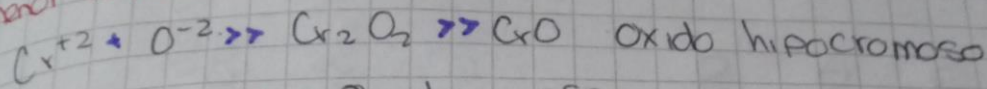


mayor Valencia Oxido...ico

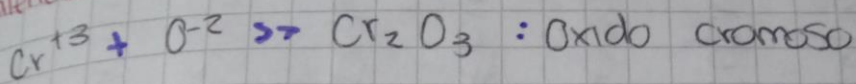


Tres valencias

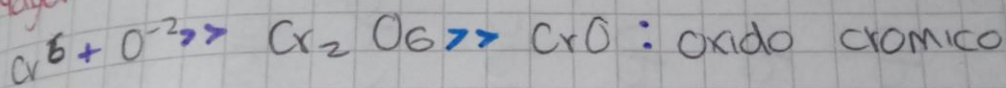
menor valencia Oxido... hipo...oso



Valencia intermedia Oxido oso

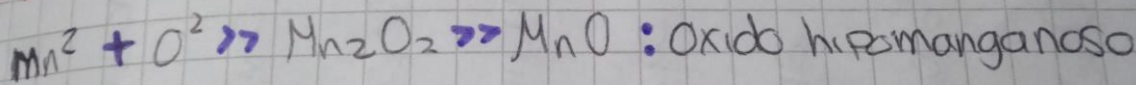


Mayor valencia Oxido ico

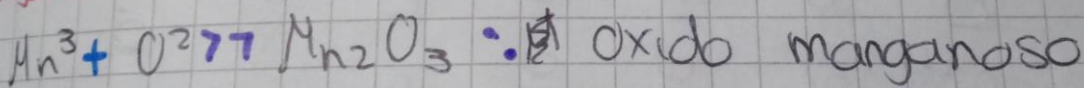


Cuatro valencias

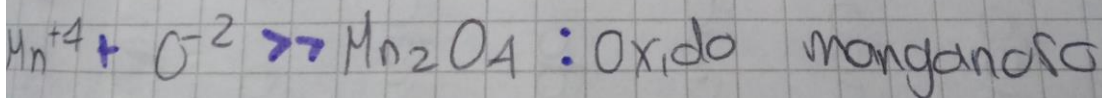
Primera valencia (baja) oxido hipo...oso



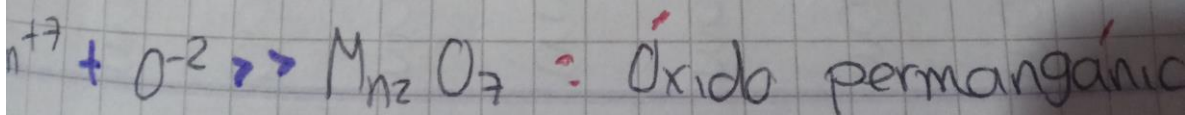
Segunda valencia Oxido...oso



Tercera valencia Oxido ico



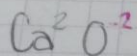
Cuarta valencia Oxido Perico



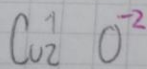
Formula	SISTEMATICA''	TRADICIONAL
Ca^2O^{-2}	monóxido de Calcio	Oxido Calcico
Cu^1O^{-2}	monóxido de cobre	Oxido cuproso
Cu^2O^{-2}	monóxido de cobre	Oxido coprico
Fe^2O^{-2}	monóxido de Hierro	Oxido ferroso
$\text{Fe}^3\text{O}_{\frac{2}{3}}$	trioxido de dihierro	Oxido ferrico
$\text{K}^1\text{O}_2^{-2}$	monóxido de Dipotasio	Oxido potasico
Sn^4O_2	Dioxido de Estano	oxido estainico
$\text{Ag}^1\text{O}_2^{-2}$	monóxido de Diplata	oxido platino

Formula

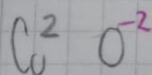
Stock



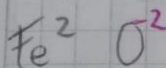
Oxido de Calcio (II)



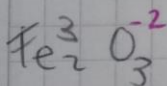
Oxido de cobre (I)



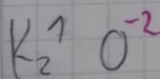
Oxido de cobre (II)



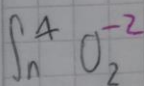
Oxido de Hierro (II)



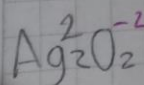
Oxido de Hierro (III)



Oxido de Potasio (I)



Oxido de Estaño (IV)



Oxido de Plata (I)