

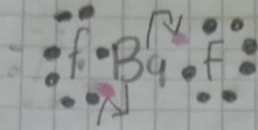
Actividad.

Graticar los siguiente compuestos con la estructura de Lewis y electronegatividades.

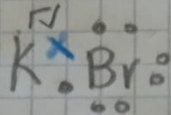
BaF ₂	Ba (metal)	F (no metal)
KBr	K (metal)	Br (no metal)
FeI ₃	Fe (metal)	I (no metal)
AlI ₃	Al (metal)	I (no metal)
CaBr ₂	Ca (metal)	Br (no metal)

Solución

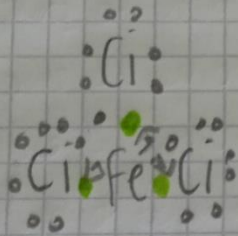
$$\text{BaF}_2 = 3.98 - 0.89 = 3.09$$



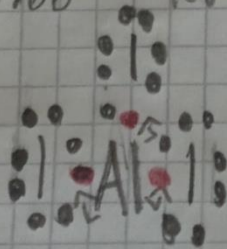
$$\text{KBr} = 2.96 - 0.82 = 2.14$$



$$\text{FeCl}_3 = 3.16 - 1.83 = 1.33$$



$$\text{AlI}_3 = 2.66 - 1.61 = 1.09$$



$$\text{CaBr}_2 = 2.96 - 1.00 = 1.96$$

