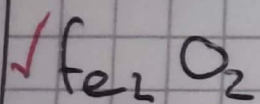


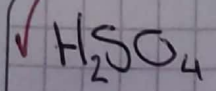
$$\text{C} = 12 \times 1 = 12 \quad = 44 \text{ uma}$$

$$\text{O} = 16 \times 2 = 32$$



$$\text{Fe} = 56 \times 2 = 112 \quad = 160 \text{ uma}$$

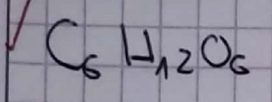
$$\text{O} = 16 \times 3 = 48$$



$H = 1 \times 2 = 2 \quad = 98 \text{ uma}$

$S = 32 \times 1 = 32$

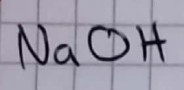
$O = 16 \times 4 = 64$



$C = 12 \times 6 = 72 \quad = 180 \text{ uma}$

$H = 1 \times 12 = 12$

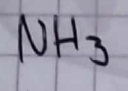
$O = 16 \times 6 = 96$



$Na = 23 \times 1 = 23 \quad = 40 \text{ uma}$

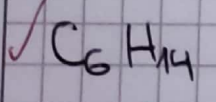
$O = 16 \times 1 = 16$

$H = 1 \times 1 = 1$



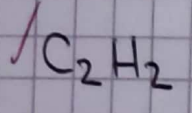
$N = 14 \times 1 = 14 \quad = 17 \text{ uma}$

$H = 1 \times 3 = 3$



$C = 12 \times 6 = 72 \quad = 86 \text{ Uma}$

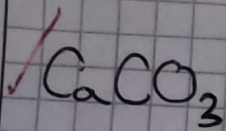
$H = 14 \times 1 = 14$



$C = 12 \times 2 = 24 \quad = 26 \text{ Uma}$

$H = 1 \times 2 = 2$

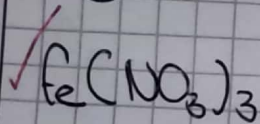
Masa Molecular



$$\text{Ca} = 40 \times 1 = 40$$

$$\text{C} = 12 \times 1 = 12 \quad = 100 \text{ uma}$$

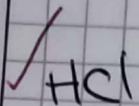
$$\text{O} = 16 \times 3 = 48$$



$$\text{Fe} = 56 \times 1 = 56$$

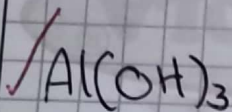
$$\text{N} = 14 \times 3 = 42 \quad = 258 \text{ uma}$$

$$\text{O} = 16 \times 9 = 144$$



$$\text{H} = 1 \times 1 = 1 \quad = 36 \text{ uma}$$

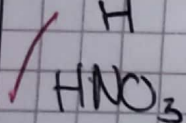
$$\text{Cl} = 35 \times 1 = 35$$



$$\text{Al} = 27 \times 1 = 27 \quad = 132 \text{ uma}$$

$$\text{O} = 16 \times 3 = 48$$

$$\text{H} = 1 \times 3 = 3$$



$$\text{H} = 1 \times 1 = 1 \quad = 63 \text{ uma}$$

$$\text{N} = 14 \times 1 = 14$$

$$\text{O} = 16 \times 3 = 48$$