Molaridad HNO_3

1. $P = 95g$
$V = 25g$ 0.025

N $1 \times 1 = 1$
H $1 \times 1 = 1$
N $1 \times 14 = 14$
O $3 \times 16 = 48$
$\overline{Pm = 63}$

$P = \dfrac{95}{(63)(0.025)} \; n = 60.3$

2. CH_3O_4

C $1 \times 12 = 12$
H $21 \times 1 = 4$
O $1 \times 16 = 16$
$\overline{Bn \; 32}$

$(32)(6.03)$
$\overline{}$
35

$m = 21.8$