

Actividad

1.

Soluto = 13g
Solución = 110g%

$$\frac{13}{110} = 0.118 \times 100 = 11.8\%$$

2.

Soluto = 10g
Solución = 123g%

$$\frac{10}{123} = 0.081 \times 100 = 8.1\%$$

3. Solvente = 15 ml
Soluto = 2.04 ml
Solución = 17.04 ml

$$\frac{15}{17.04} = 0.119 \times 100 = 11.9\%$$

4. Solvente 11 ml
Soluto 2.09 ml
Solución 13.09 ml

$$\frac{2.4}{13.09} = 0.159 \times 100 = 15.9\%$$

Soluto = 55 gr

Solvente = 239 gr

o/o

S = 55 gr

D = 239 gr

$$\frac{M}{M} \% = \frac{55}{239} = 0.18 \times 100 = 18\%$$

Solución

SO = 294 gr

$$\frac{55 \text{ gr}}{294 \text{ gr}} \times 100$$

Soluto = 2.3 kg

Solvente = 6L

S = 2.3 kg

D = 6L

SO = 8.3L

$$\% = \frac{2.3}{8.3} \times 100 = 27\%$$

o/o

Soluto 55 gr

Solvente 239 gr

Solución = 294 gr

$$\frac{55 \text{ gr}}{294 \text{ gr}} \times 100$$

$$\frac{55 \text{ gr}}{294 \text{ gr}} \times 100$$

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