

$$H = 1 \quad O = 16$$

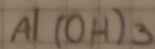
$$v = 49,04 \text{ g}$$

$$m = 250 \text{ ml}$$

$$M = \frac{250}{4,384} = 0,057$$

$$\text{COH} \quad m = 4,50$$

$$v = 950 \text{ ml} / 1000 = 0,95 \text{ L}$$



$$P_m = H = 3 \times 1 = 3$$

$$O = 3 \times 16 = 48$$

$$\text{Al} \quad 1 \times 27 = \frac{27}{27 \text{ mol}}$$

$$N = \frac{(9,50) (3)}{(27) (0,95)}$$

$$N = \frac{28,5}{55,1} = 0,51$$

