

### Actividad

1o Obten la normalidad de  $\text{Al}(\text{OH})_3$  con 450 ml y 9,50 g

$$M = 9,50 \text{ g}$$

$$V = 450 \text{ ml} / 100 = 0,45 \text{ L}$$

$$\text{Al} \times 1 = 27$$

$$\text{O} \times 3 = 48$$

$$\text{H} \times 3 = 3$$

$$\text{PM} = 78 \text{ g/mol}$$

$$\frac{9,50 \times 3}{78 \times 0,45}$$

$$\rightarrow \frac{28,5}{35,1} = 0,811$$

$$N = 0,811$$