

⑤  $\text{NH}_3$

$$\text{N}_{1 \times 14} = 14$$

$$\text{H}_{3 \times 1} = 3$$

①  $\text{C}_2\text{H}_6$

$$\text{C}_{2 \times 12} = 24$$

$$\text{H}_{6 \times 1} = 6$$

④  $\text{C}_2\text{H}_2$

$$\text{C}_{2 \times 12} = 24$$

$$\text{H}_{2 \times 1} = 2$$

②  $\text{CO}_2$

$$\text{C}_{1 \times 12} = 12$$

$$\text{O}_{2 \times 16} = 32$$

③  $\text{Fe}_2\text{O}_3$

$$\text{Fe}_{2 \times 56} = 112$$

$$\text{O}_{3 \times 16} = 48$$



$$\begin{aligned}\text{Ca}_{11} &= 40 \\ \text{C}_{11} &= 12 \\ \text{O}_{33} &= 48\end{aligned}$$



$$\begin{aligned}\text{H}_{11} &= 1 \\ \text{Cl}_{11} &= 35\end{aligned}$$



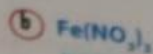
$$\begin{aligned}\text{H}_{11} &= 1 \\ \text{N}_{11} &= 14 \\ \text{O}_{33} &= 48\end{aligned}$$



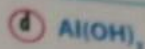
$$\begin{aligned}\text{C}_{26} &= 72 \\ \text{H}_{12} &= 12 \\ \text{O}_{26} &= 48\end{aligned}$$



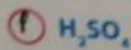
$$\begin{aligned}\text{Mg}_{11} &= 24 \\ \text{O}_{11} &= 16\end{aligned}$$



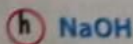
$$\begin{aligned}\text{Fe}_{11} &= 56 \\ \text{N}_{13} &= 42 \\ \text{O}_{36} &= 96\end{aligned}$$



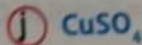
$$\begin{aligned}\text{Al}_{11} &= 27 \\ \text{O}_{12} &= 48 \\ \text{H}_{12} &= 12\end{aligned}$$



$$\begin{aligned}\text{H}_{22} &= 2 \\ \text{S}_{11} &= 32 \\ \text{O}_{44} &= 64\end{aligned}$$



$$\begin{aligned}\text{Na}_{11} &= 23 \\ \text{O}_{11} &= 16 \\ \text{H}_{11} &= 1\end{aligned}$$



$$\begin{aligned}\text{Cu}_{11} &= 63 \\ \text{S}_{11} &= 32 \\ \text{O}_{44} &= 64\end{aligned}$$

