las siguientes moléculas.

a) $CaCO_3$

Ca = 40 × 1 = 40
C = 38 × 3 = 174

b) $Fe(NO_3)_3$

Fe = 56 × 1 = 56
N = 14 × 6 = 42
O = 16 × 6 = 96

c) HCl

H = 1 × 1 = 1
Cl = 35 × 1 = 35

d) $Al(OH)_3$

Al = 27 × 1 = 27
O = 16 × 3 = 48
H = 1 × 3 = 3

e) HNO_3

H = 1 × 1 = 1
N = 14 × 1 = 14
O = 16 × 3 = 48

f) H_2SO_4

H = 1 × 2 = 2
32 × 1 = 32
O = 16 × 4 = 64

g) $C_6H_{12}O_6$

C = 12 × 6 = 72
H = 1 × 12 = 12
O = 16 × 6 = 96

h) $NaOH$

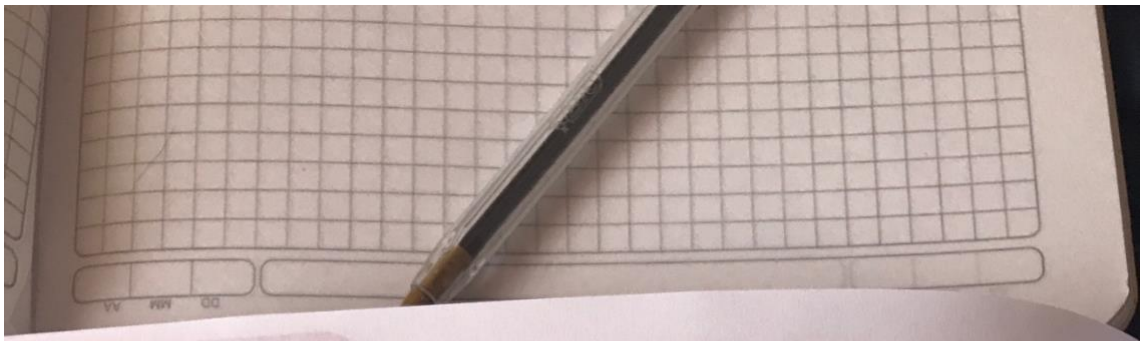
Na = 23 × 1 = 23
O = 16 × 1 = 16
H = 1 × 1 = 1

i) MgO

Mg = 24 × 1 = 24
O = 16 × 1 = 16

j) $CuSO_4$

Cu = 63 × 1 = 63
S = 32 × 1 = 32
O = 16 × 4 = 64



k NH_3

N = 14 × 1 = 14
H = 1 × 3 = 3

l C_6H_{14}

C = 12 × 6 = 72
H = 1 × 14 = 14

m C_2H_2

C = 12 × 2 = 24
H = 1 × 2 = 2

n CO_2

C = 12 × 1 = 12
O = 16 × 2 = 32

o Fe_2O_2

Fe = 56 × 2 = 112
O = 16 × 2 = 32

Cálculo de masa

Para realizar cálculos